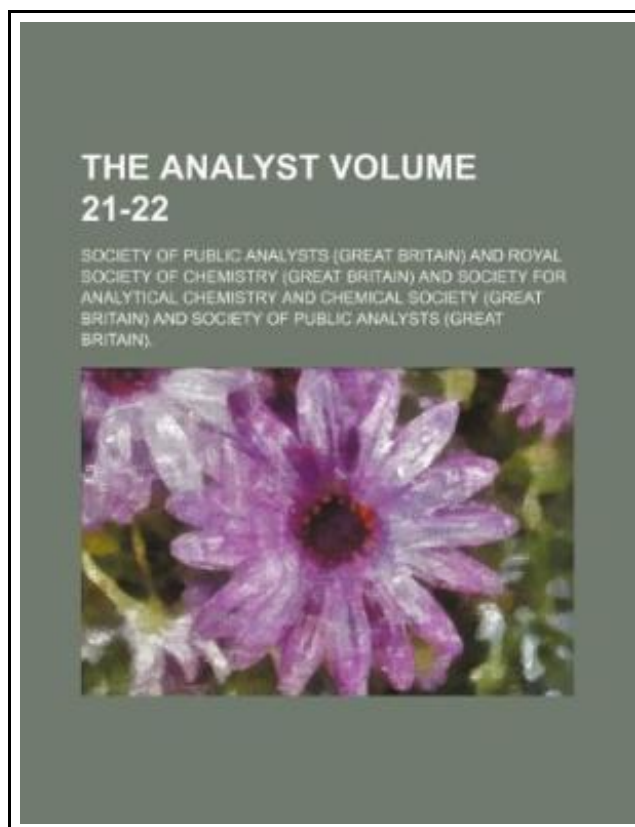


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1896 Excerpt: .other hand, the method ought to do good service, as it may safely be assumed that its solubility is far less than that of stearic acid. We hope to recur to this subject at some future time. To summarize in a few words the method which we recommend: Prepare a supply of alcohol saturated at 0 C. with pure stearic acid, or with stearic acid which only contains traces of palmitic acid. Dissolve from 0.5 to 1 gramme of the mixture of the fatty acids to be examined, if these are solid, or about 5 grammes if fluid, in about 100 c.c. (exact measurement is not necessary) of the stearic alcohol solution. Leave in an ice-bath overnight, agitate the mixture next morning, and allow to stand in ice for a short time; filter off while the mixture remains in ice, wash with stearic alcohol at 0 C, dry and weigh. Determine the melting-point of the product, which should not be much less than 68.5 C. Discussion. Mr. Allen said that the subject was one upon which only limited information existed, and he congratulated the authors upon having made a very great advance. It was only in recent years that anything was possible in the way of quantitative differentiation of the fatty acids. Something could be done with the aid of the iodine absorption, but, as had been pointed out in the paper, this had to be taken as representing the proportions in which the saturated...



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